(6 pa			3.	Wha	at is the relatio	n hetweer	kinetic ene	rgy (E) of	
	Reg. N	Vo.:	ο,		s and its pressu			16) (11) 01	
Code No.: 10435 E Sub. Code: CMPH 41		Sub. Code: CMPH 41		(a)	P = 2/3 E	(b)	P = 3 E		
				(c)	P = 1/3 E	(d)	E = 2/3 P		
B.Sc. (CBCS) DEGREE EXAMINATION, APRIL 2023. Fourth Semester			4.	Vander waals equation help us accurately define the physical state of a real					
	Physics	s – Core		(a)	Gas	(b)	Solid		
	HEAT AND THERMODYNAMICS			(c)	Liquid	(d)	Semi solid		
(For those who joined in July 2021 onwards)									
Time: Three hours Maximum: 75 marks		5.	its ability to conduct				leasure or		
	PART A — (10	\times 1 = 10 marks)		(a)	Heat	(b)	Volume		
	Answer AL	L questions.		(c)	Temperature	(d)	Entropy		
	Choose the correct answer:		6.	6. A black body is an idealized physical body that			body that		
1.	The basic principle of the joule Thomson effect is based on the transfer of heat		o.	absorbs all incident electromagnetic					
						(h)	emit		
	(a) Heat	(b) Temperature		(a)	absorbs	(b)	conduct		
	(c) Volume	(d) Entropy		(c)	reflect	(d)			
2.	The lowest pressure	lambda point at 2.172 K is	7.		First law of thermodynamics also known as the law of conservation of				
	(a) 0.0497 atm	(b) 0.003 atm		(a)	energy	(b)	momentu	n	
	(c) 0.0457 atm	(d) 0.0476 atm		(c)	mass	(d) Page 2	density Code No.		
8.	Carnot cycle has efficiency in all		1	PART B — $(5 \times 5 = 25 \text{ marks})$					
	cycle.			Answer ALL questions, choosing either (a) or (b).					
	(a) Manimum		B	Ea	ch answer shou	ıld not exc	eed 250 wor	ds.	
	(a) Maximum		11.	(a)	State and exp	lain Joule	s Kelvin effe	ect.	
	(b) Minimum					Or			
	(c) Both minimum a	nd maximum		(b)	To write a	ny five	applications	of low	
	(d) None	**	12.		temperature.				
9.	The third law of thermodynamics states that the			(a)	Explain abou	t kinetic t	heory of gas		
	entropy of pure crystal at absolute zero is					\mathbf{Or}			
		* * * * * * * * * * * * * * * * * * *		(b)	State and det	termine W	onder walls	constant.	
	(a) zero	(b) low	13.	(a)	State and exp	plain ther	mal conduct	vity.	
	(c) high	(d) medium				Or			
	7, 7			(b)	To Describe	about blac	k body radia	tion.	
10.	The behavior of carnot engine best understand by using diagram.		14.	(a)	State and thermodynam		n zeroth	law of	
	(a) TS	(b) VS				\mathbf{Or}	1		
	(c) HT	(d) V-I		(b)	What are the		ce between o	otto engine	

Page 3 Code No.: 10435 E

Page 4 Code No.: 10435 E

15. (a) How to draw the temperature entropy diagram?

Or

(b) State and explain third law of thermodynamics.

PART C — $(5 \times 8 = 40 \text{ marks})$

Answer ALL questions, choosing either (a) or (b).

Each answer should not exceed 600 words.

16. (a) What is the difference between liquefaction hydrogen and helium gas?

Or

- (b) Describe the process of production low temperature.
- (a) Briefly explain maxwells law of distribution of molecular velocities.

Or

- (b) To write relation between the wander walls constant critical constant.
- 18. (a) To give the detail explanation about thermal conductivity of good conductor.

Or

(b) Explain construction and working of Lee's disc method with suitable diagram.

Page 5 Code No.: 10435 E

 (a) Explain work done during the adiabatic and isothermal process.

Or

- (b) Explain construction and working of carnot engine with suitable diagram.
- 20. (a) What are the effect of pressure on melting and boiling point?

Or

(b) To derive the expression of maxwells thermodynamics relation and its application.

Page 6 Code No.: 10435 E