(8	pages)	

Reg.	Ma	2	
neg.	NO.	1	

Code No.: 10320 E Sub. Code: AMCH 41

B.Sc. (CBCS) DEGREE EXAMINATION, APRIL 2023

Fourth Semester

Chemistry - Core

ORGANIC CHEMISTRY — II

(For those who joined in July 2020 only)

Time: Three hours

Maximum: 75 marks

PART A — $(10 \times 1 = 10 \text{ marks})$

Answer ALL questions.

Choose the correct answer:

- The formation of cyanohydrin from a ketone is an example of ——— reaction.
 - (a) electrophillic addition
 - (b) nucleophillic addition
 - (c) nucleophillic substitution
 - (d) electrophillic substitution

- Under Wolff Kinhnor reduction conditions, the conversions which may be brought about are
 - (a) Cyclohexanone into cyclohexons
 - (b) Benzaldehyde into benzyl alcohol
 - (c) Cyclohexanone into cyclohexanol
 - (d) None
- 3. Which one of the following is the most strong acid?
 - (a) Acetic acid
- (b) n-butyric acid
- (c) Propionic acid
- (d) Formic acid

4.
$$CH_2 - COOH$$

$$CH_2 \cdot COOH$$

$$CH_2 \cdot COOH$$

$$CH_2 \cdot COOH$$

Succinic acid X is -

$$\begin{array}{cccc} & CH_2-CH_3 & CH_2-COOH \\ \text{(a)} & | & \text{(b)} & | \\ & CH_2-CH_3 & CH_2-CH_3 \end{array}$$

(c)
$$| CH_2 - CO |$$
 (d) $| COOH |$ COOH

Page 2 Code No.: 10320 E

- Ethyl bromide reacts with sodium hydrosulphide gives ———
 - (a) ethanethiol
- (b) diethyl sulphide

Page 3 Code No.: 10320 E

- (c) mustard gas
- (d) sulphonal
- 6. The metal used in Reformutnky reaction is -
 - (a) Mg
- (b) Li
- (c) Pb
- (d) Zn
- 7. Nitro acinitro tautomerism is not exhibited by
 - (a) primary nitro compound
 - (b) tertiary nitro compound
 - (c) secondary nitro compound
 - (d) none of the above
- 8. The hydrogen atom in the methylene group is strongly acidic because of ——— groups.
 - (a) electron with drawing
 - (b) electron releasing
 - (c) both (a) and (b)
 - (d) none of the above

- 9. Which of the following correctly lists the conformations of cyclohexane in order of increasing energy?
 - (a) Chair < twist < boat < half chair
 - (b) Chair < boat < twist < half chair
 - (c) Half chair < boat < twist < chair
 - d) Chair < twist < half chair < boat
- 10. Which statement about cyclohexane is incorrect?
 - (a) Each carbon atom is sp³ hybridient
 - (b) The cyclohexane ring can flip between chair and bond conformern
 - (c) Cyclokexane suffers ring strain
 - (d) Hydrogen atom occupy equitorial or axial sites

PART B — $(5 \times 5 = 25 \text{ marks})$.

Answer ALL questions choosing either (a) or (b). Each answer should not exceed 250 words.

 (a) Discuss the mechanism of nucleophillic addition of grignard reagent with acetaldehyde.

Or

(b) What is Meerwein - Pondort - verley reduction? Explain with example.

Page 4 Code No.: 10320 E

12. (a) How is urea prepared? How does it react with (i) HNO₂ (ii) NH₂NH₂.

Or

- (b) Discuss the effect of substituents on the acidily of carboxylic acid with suitable example.
- (a) How is diethylsulphide prepared? How is it react with (i) H₂O₂ (ii) Ni.

Or

- (b) How is methyllithium react with the following (i) HCHO (ii) CH₃CHO (iii) CH₃COCH₃.
- 14. (a) How are the following prepared from ethyl acetoacetate?
 - (i) Butanoic acid
 - (ii) Crotonoic acid.

Or

(b) Discuss the mechanism of nitroso - oxime tantomerism what are the evidences for the two forms?

Page 5 Code No.: 10320 E

- 18. (a) How are the following prepared from diethylzinc?
 - (i) Primary alcohol
 - (ii) · Secondary alcohol
 - (iii) Tertiary alcohol
 - (iv) Keton.

Or

- (b) Discuss the preparation and properties of
 - (i) Sulphone
 - (ii) Sulphonal
 - (iii) Mustard gas.
- (a) Explain the following tantomerisms and the evidences for them
 - (i) Keto enol
 - (ii) Nitro acinitro.

O

- (b) How are the following prepared from diethyl mulonate?
 - (i) Succinic acid
 - (ii) Aceto acetic acid
 - (iii) Poutanoic acid.

15. (a) Explain Sachse - Mohr theory of strainless ring.

Or

(b) How is the civatone prepared? Write its structure.

PART C — $(5 \times 8 = 40 \text{ marks})$

Answer ALL questions choosing either (a) or (b). Each answer should not exceed 600 words.

- (a) (i) Explain the relative reactivities of carboxyl compounds.
 - (ii) Discuss the mechanism of Knoevenagal reaction.

Or

- (b) Give the preparation, properties and uses of chloral.
- 17. (a) Explain the reaction of action of heat on
 - (i) hydroxy acids
 - (ii) dicarboxylic acid.

Or

- (b) (i) Discuss the mechanism of esterification.
 - (ii) Write note on the optical property of lactic acid.

Page 6 Code No.: 10320 E

20. (a) Discuss the conformational analysis of mononutrituted cyclohexane.

Or

- (b) Write note on:
 - (i) Baeyers strain theory
 - (ii) Coulson Moffit's concept.

Page 7 Code No.: 10320 E