Reg.	No.	:	

Code No.: 5409

Sub. Code: ZCHE 11

M.Sc. (CBCS) DEGREE EXAMINATION, APRIL 2023

First Semester

Chemistry

Elective-GREEN CHEMISTRY-TECHNIQUES AND APPLICATIONS

(For those who joined in July 2021 onwards)

Time: Three hours

Maximum: 75 marks

PART A - (10 × 1 = 10 marks)

Answer ALL questions.

Choose the correct answer:

- 1. Aim of Green chemistry is
 - (a) Design chemical products and process that maximize profits
 - (b) Design safer chemical products and processes that reduce or eliminate the use and generation of hazardous substances
 - (c) Design chemical products and processes that work most efficiently
 - (d) Utilize non-renewable energy
- 6. What is the solvent used for Diels-Alder reaction in Green synthesis?
 - (a) Formaldehyde (b)
- (b) Ethanol
 - (c) Water
- (d) Benzene
- Photolysis of benzophenone (in sunlight) in presence of an alcoholic solvent (preferably isopropyl alcohol) gives———
 - (a) Benzopinacol
 - (b) Benzoic acid
 - (c) Benzyl alcohol
 - (d) Phenol
- 8. Which of the following is used as a Friedel-Crafts catalyst?
 - (a) Anhydrous Aluminium chloride
 - (b) Aluminium Flouride
 - (c) Tungsten
 - (d) Activated charcoal
- 9. Which among the following have a large amount of installed grid interactive renewable power capacity in India?
 - (a) Biomass power
 - (b) Wind power
 - (c) Solar power
 - (d) Small Hydro power
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- Green chemistry reduces the use of
 - (a) Energy
- (b) Liquid fuels
- (c) Gaseous fuels
- (d) Solid fuels
- 3. Which one is not the advantage of zeollte catalysed Friedalcrafts acylation?
 - (a) No solvent is required
 - (b) 0.035 kg aqueous effluent per kg instead of 4.5 kg aqueous effluent per kg
 - (c) Greater than 95% yield of higher purity instead of 85- 95% yield in traditional method
 - (d) Hydrolysis of products and phase separation is necessary
- 4. Name the enzyme which catalyzes the oxidationreduction reaction?
 - (a) Transaminase
 - (b) Glutamine synthetase
 - (c) Phosphofructokinase
 - (d) Oxidoreductase
- Universal solvent is
 - (a) Formaldehyde (b) Water
 - (c) Ethanol
- (d) Benzene

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- 10. Which of the following is not a disadvantage of Hydro Power?
 - (a) A large land area submerges into the water leading to deforestation
 - (b) It causes dislocation of a large population and rehabilitation
 - (c) It causes ecological disturbances like earthquake
 - (d) It causes calamities like landslides

PART B — $(5 \times 5 = 25 \text{ marks})$

Answer ALL questions, choosing either (a) or (b).

- 11. (a) (i) How can we reduce non-renewable resources?
 - (ii) How do you calculate mass intensity?

Or

- (b) (i) How do you calculate E factor in green chemistry?
 - (ii) What does a low E factor Mean?

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- 12. (a) (i) What are the advantages of green catalysts?
 - (ii) Why enzyme is a green catalyst?

Or

- (b) What is bio-catalysis? Discuss the role of enzymes in catalytic oxidation and catalytic reduction.
- 13. (a) Explain Aqueous Phase Diels Alder reaction

Or

- (b) Write notes on Claisen rearrangement.
- 14. (a) Give a brief note Photoreduction reactions.

Or

- (b) (i) What is Strecker synthesis?
 - (ii) Discuss the mechanism of Strecker synthesis
- 15. (a) Write notes on renewable energy sources.

Or

(b) Give a brief account of hydroelectric and geothermal Power.

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 (a) Discuss the effects of solvents on microwave assisted synthesis.

Or

- (b) (i) What do you mean by sonochemistry?
 - (ii) Discuss the Principle of sonochemistry
 - (iii) Write notes on ultra sound assisted Cannizzaro reaction
- 20. (a) Explain the Principle, Types and Applications of Fuel Cells.

Or

- (b) Write note on
 - (i) Hydroelectic Power
 - (ii) Biomass
 - (iii) Wind Power and
 - (iv) Geothermal power

PART C — $(5 \times 8 = 40 \text{ marks})$

Answer ALL questions, choosing either (a) or (b).

 (a) Explain the twelve Principles of Green Chemistry.

Or

- (b) Write notes on alternate energy sources and efficient energy improvements.
- 17. (a) Discuss the applications of zeolites, silica, alumina and clay catalysts in green chemical reactions.

Or

- (b) Discuss the applications of crown ethers in oxidation, substitutions, elimination and esterification reactions.
- 18. (a) Explain the role of solvents in synthesis.

Or

(b) Write any two reactions using super critical carbon dioxide and two reactions using super critical water.

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